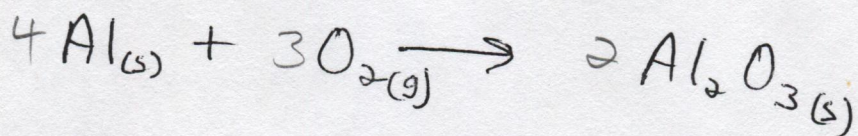


Chemistry I Quiz (B) Teacher
Chapter 11

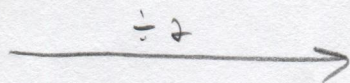
How much ~~is~~ Aluminum oxide will be produced by the complete oxidation of 126g of Aluminum. The reaction is



126g

↓ $\div 26.98 \text{g/mol}$

4.67 mol



2.335 mol

↓ $\times 101.96 \text{g/mol}$

238g